

Notiz / Note

Relationship between Dominant Conformation and Diastereoselectivity of *O*-Acetal Formation of Saturated 2*H*-3,1-Benzothiazine-2-thionesPál Perjési*^a and Gyula Batta^bDepartment of Medical Chemistry, University Medical School^a,
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Stereospecific formation of *O*-acetals **4**, **5** of the isomeric octahydro-8a-hydroxy-4-phenyl-2*H*-3,1-benzothiazine-2-thiones **1–3** is reported. The stereoselectivity of the reaction is deter-

mined by the dominant conformations. During all reactions the phenyl substituent retains its equatorial position.

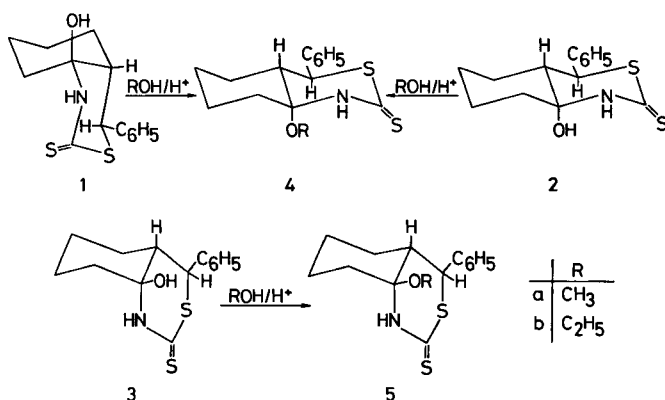
In a preceding paper we have reported that (*E*)-2-benzylidenecyclohexanone reacts with dithiocarbamic acid to yield isomeric octahydro-8a-hydroxy-4-phenyl-2*H*-3,1-benzothiazine-2-thiones (**1–3**)^[1]. Compound **1** has been found to be an unstable isomer which undergoes thermal or acid-catalyzed isomerization to **2**^[1]. Treatment of **1–3** with *p*-H₃CC₆H₄SO₃H, CF₃CO₂H, or (C₂H₅)₂O · BF₃ in aprotic solvents affords isomeric dehydration products, depending on the configuration of **1–3** and the dehydration method used^[2]. In methanol or ethanol solutions of **1–3** in the presence of a catalytic amount of hydrochloric acid, however, the formation of the *O*-alkyl derivatives **4** and **5** is observed (Scheme)^[3].

an unsubstituted N-1 atom^[7]. In the case of **1** the N-outside conformation predominates due to the preference of the bulky 4-phenyl group to adopt the energetically more preferred equatorial position^[2]. Similarly, the predominance of the N-outside conformation with an equatorial 2-aryl group has been reported for one of the epimeric saturated 2-(4-nitrophenyl)-3,1-benzoxazines^[8].

This preference of the 4-phenyl group for the equatorial orientation may be responsible for the stereospecific *O*-acetal formation of **2** and **3** as well. On the one hand, inversion of configuration of the C-8a chiral center of **2** would result in the formation of the *O*-alkyl derivatives of **1**, which are not formed from **1** due to its epimerization mentioned above. On the other hand, inversion of configuration at C-8a of **3** would result in the formation of the *trans*-annulated isomer with an energetically non-preferred axial 4-phenyl substituent. The formation of an adduct with such a configuration has not been observed even in the reaction of 2-benzylidenecyclohexanone with dithiocarbamic acid^[1,2], most probably for the same reason.

The *O*-acetal structure of **4** and **5** is evidenced by their IR and NMR spectra (see Experimental and Table 1). The characteristic ¹H- and ¹³C-NMR spectroscopic data of **4** and **5** are very similar to those of **2** and **3**, respectively, suggesting the same configuration and conformation of the corresponding compounds (see Scheme). This has further been supported by ¹H-¹H NOE experiments with **4** and **5**, the results of which are presented in Table 2.

Scheme



The formation of the *O*-alkyl derivatives of **2** and **3** has been found to proceed stereospecifically. ¹H-NMR analysis^[4] of the crude products obtained in the reactions of **2** and **3** shows only the presence of **4** and **5**, respectively. Treatment of a mixture of **1** and **2**, however, results in the exclusive formation of **4** under the same conditions.

This latter observation is in accordance with the pronounced tendency of **1** to undergo isomerization to **2**, which can be well interpreted as a result of its *N*-outside conformation, which is the less preferred one with this type of saturated heterocycles containing

Experimental

Melting points are uncorrected: Boetius apparatus. — IR: Spcorder 75 IR. — ¹H and ¹³C NMR: TMS as internal standard, Perkin-Elmer R-12 (60 MHz) and Bruker WP-200 SY (200.13 and 50.3 MHz, respectively). COSY-60 ¹H-¹H correlation spectra, long-range ¹H-¹³C connectivities, and steady-state ¹H-¹H NOE measurements were performed as described earlier^[9]. — Elemental analyses: Department of Organic Chemistry, Eötvös József University, Budapest.

General Procedure for the Formation of 4 and 5: 1.0 mmol of **1–3** (**2**, **3**, or approximately a 1:1 mixture of **1** and **2**) is dissolved in 100 ml of anhydrous methanol or ethanol, and the mixture is gently

Table 1. ^{13}C -NMR (50.3 MHz) spectra of compounds 1–5

Compounds	C-2	C-1'	C-2'	C-3'	C-4'	C-8a	C-4	C-4a	C-8	C-7	C-6	C-5	Others
1[a,b]	189.96	137.19	128.89[c]	128.27[c]	128.00	85.34	48.27	42.90	34.11	18.93	21.85	23.59	
2[a]	191.97	137.25	128.65		127.81	80.92	49.06	43.55	36.87	20.99	24.57	25.44	
3[a]	194.02	136.79	128.57[c]	128.28[c]	127.74	82.26	46.66	43.99	37.73	21.48	22.45	23.78	
4a[a]	193.76	137.22	128.74		127.99	84.11	49.03	43.50	30.57	20.72	24.40	25.66	48.04[e]
4b[a]	193.56	137.28	128.74		127.96	83.94	49.10	43.58	31.44	20.83	24.46	25.66	55.62[f], 15.37[g]
5a[d]	195.98	136.30	128.08[c]	128.15[c]	127.42	85.53	46.46	42.89	32.03	21.92	23.71	21.25	48.14[e]
5b[d]	195.80	136.38	128.17[c]	128.10[c]	127.43	85.39	46.49	42.96	32.92	21.99	23.71	21.16	55.71[f], 15.38[g]

[a] $[\text{D}_6]\text{DMSO}$. — [b] Determined from the mixture of 1 and 2. — [c] Interchangeable assignments. — [d] Polysol. — [e] OCH_3 . — [f] $\text{CH}_2(\text{OEt})$. — [g] $\text{CH}_3(\text{OEt})$.

Table 2. Results of ^1H - ^1H 1D-NOE experiments of compounds 4 and 5

Compound	Irradiated proton (d)	Observed proton NOE (%)
4a	4a-H (2.20) 8- H_{eq} (2.48) OCH_3 (3.23) 4-H (4.42) ArH NH (10.89)	NH (1.6), ArH (18.4), 4-H (2.0); NH (6.0), OCH_3 (7.5), 8- H_{ax} (25.0); NH (3.8), 8- H_{eq} (5.0); ArH (20.0); 4-H (16.4), 4a-H (7.9); 8- H_{eq} (4.3)
	4a-H (2.17) 8- H_{eq} (2.47) CH ₂ (OEt) (3.60) 4-H (4.42) ArH NH (10.87)	NH (1.2), ArH (16.0), 4-H (2.4) NH (6.0), CH ₂ (OEt) (ca.5), 8- H_{ax} (ca.12) NH (4.5), 8- H_{eq} (3.3), CH ₃ (OEt) (12.3); ArH (17.0); 4-H (14.2), 4a-H (7.6); 8- H_{eq} (4.1)
5a	4a-H (1.99) 8- H_{eq} (2.57) OCH_3 (3.31) 4-H (5.22) ArH NH (10.61)	ArH (4.3), 4-H (15.2), [8- H_{ax} + 6- H_{ax}] (4.5); NH (8.0), 8- H_{ax} (31); NH (4.2), 4-H (2.1), 8- H_{eq} (3.2); ArH (19.1), 4a-H (8.4); 4-H (19.1), 4a-H (2.0); 8- H_{eq} (6.1)
	4a-H (1.98) 8- H_{eq} (2.56) CH ₂ (OEt) (3.60) 4-H (5.27) ArH NH (10.77)	ArH (5.0), 4-H (13.9), 8- H_{ax} (3.4); NH (5.9), CH ₂ (OEt) (7.4), 8- H_{ax} (21.7); NH (4.0), 4-H (1.2), 8- H_{eq} (4.1), CH ₃ (OEt) (12.5); ArH (18.4), 4a-H (8.9), CH ₃ (OEt) (1.1); 4-H (17.3), 4a-H (2.4); CH ₂ (OEt) (6.1), 8- H_{eq} (4.5)

heated. The clear solution is then cooled to room temp., and 0.1 ml of a saturated methanolic or ethanolic HCl solution is added. The reaction mixture is kept at room temp. for 2 d, and the solvent is removed under reduced pressure to yield colorless crystals of 4 and 5.

(4 α ,4 $\alpha\alpha$,8 $\alpha\beta$)-1,4,4a,5,6,7,8,8a-Octahydro-8a-methoxy-4-phenyl-2H-3,1-benzothiazine-2-thione (4a): Yield: 0.27 g (92%), m.p. 185–188 °C (methanol). — IR (KBr): $\tilde{\nu}$ = 3140 cm^{-1} (NH), 2925 (CH), 1500 ($\text{C}=\text{C}_{\text{Ar}}$), 1335 (dithiourethane). — ^1H NMR: δ = 10.89

(s, 1H, NH), 7.44–7.25 (m, 5H, aromatic H), 4.42 (d, $^3J_{4,4a}$ = 11.3 Hz; 1H, 4-H), 3.23 (s, 3H, OCH_3), 2.48 (m, 1H, 8- H_{eq}), 2.20 (ddd, $^3J_{4a,5ax}$ = 11.5, $^3J_{4a,5eq}$ = 4.3 Hz; 1H, 4a-H), 1.54 (m, 2H, 7- H_{eq} , 5- H_{eq}), 1.50–0.95 (m, 5H, 5- H_{ax} , 6- H_{eq} , 6- H_{ax} , 7- H_{ax} , 8- H_{ax}). — $\text{C}_{15}\text{H}_{19}\text{NOS}_2$ (293.5): calcd. C 61.40, H 6.53, N 4.77; found C 61.23, H 6.44, N 4.68.

(4 α ,4 $\alpha\alpha$,8 $\alpha\beta$)-8a-Ethoxy-1,4,4a,5,6,7,8,8a-octahydro-4-phenyl-2H-3,1-benzothiazine-2-thione (4b): Yield: 0.29 g (94%), m.p. 184–187 °C (ethanol). — IR (KBr): $\tilde{\nu}$ = 3130 cm^{-1} (NH), 2930 (CH), 1505 ($\text{C}=\text{C}_{\text{Ar}}$), 1340 (dithiourethane). — ^1H NMR: δ = 10.87 (s, 1H, NH), 7.51–7.22 (m, 5H, aromatic H), 4.42 (d, $^3J_{4,4a}$ = 11.6 Hz; 1H, 4-H), 3.60 [m, 2H, CH₂ (OEt)], 2.47 (m, 1H, 8- H_{eq}), 2.17 (ddd, $^3J_{4a,5ax}$ = 11.3, $^3J_{4a,5eq}$ = 4.5 Hz; 1H, 4a-H), 1.53 (m, 2H, 7- H_{eq} , 5- H_{eq}), 1.50–1.05 (m, 5H, 5- H_{ax} , 6- H_{eq} , 6- H_{ax} , 7- H_{ax} , 8- H_{ax}), 1.17 [s, 3H, CH₃ (OEt)]. — $\text{C}_{16}\text{H}_{21}\text{NOS}_2$ (307.5): calcd. C 62.50, H 6.88, N 4.56; found C 62.37, H 6.73, N 4.51.

(4 α ,4 $\alpha\beta$,8 $\alpha\beta$)-1,4,4a,5,6,7,8,8a-Octahydro-8a-methoxy-4-phenyl-2H-3,1-benzothiazine-2-thione (5a): Yield: 0.27 g (92%), m.p. 194–197 °C (methanol). — IR (KBr): $\tilde{\nu}$ = 3100 cm^{-1} (NH), 2930 (CH), 1500 ($\text{C}=\text{C}_{\text{Ar}}$), 1365 (dithiourethane). — ^1H NMR: δ = 10.61 (s, 1H, NH), 7.30–7.50 (m, 5H, aromatic H), 5.22 (d, $^3J_{4,4a}$ = 3.65 Hz; 1H, 4-H), 3.31 (s, 3H, OCH_3), 2.57 (m, 1H, 8- H_{eq}), 1.99 (ddd, $^3J_{4a,5ax}$ = 11.0, $^3J_{4a,5eq}$ = 3.6 Hz; 1H, 4a-H), 1.67 (m, 2H, 7- H_{eq} , 5- H_{eq}), 1.45–0.99 (m, 5H, 5- H_{ax} , 6- H_{eq} , 6- H_{ax} , 7- H_{ax} , 8- H_{ax}). — $\text{C}_{15}\text{H}_{19}\text{NOS}_2$ (293.5): calcd. C 61.40, H 6.53, N 4.77; found C 61.21, H 6.55, N 4.53.

(4 α ,4 $\alpha\beta$,8 $\alpha\beta$)-8a-Ethoxy-1,4,4a,5,6,7,8,8a-octahydro-4-phenyl-2H-3,1-benzothiazine-2-thione (5b): Yield: 0.28 g (91%), m.p. 192–194 °C (ethanol). — IR (KBr): $\tilde{\nu}$ = 3115 cm^{-1} (NH), 2920 (CH), 1500 ($\text{C}=\text{C}_{\text{Ar}}$), 1370 (dithiourethane). — ^1H NMR: δ = 10.77 (s, 1H, NH), 7.43–7.25 (m, 5H, aromatic H), 5.27 (d, $^3J_{4,4a}$ = 3.65 Hz; 1H, 4-H), 3.60 [m, 2H, CH₂ (OEt)], 2.56 (m, 1H, 8- H_{eq}), 1.98 (ddd, $^3J_{4a,5ax}$ = 11.5, $^3J_{4a,5eq}$ = 3.5 Hz; 1H, 4a-H), 1.66 (m, 2H, 7- H_{eq} , 5- H_{eq}), 1.50–1.10 (m, 5H, 5- H_{ax} , 6- H_{eq} , 6- H_{ax} , 7- H_{ax} , 8- H_{ax}), 1.19 [s, 3H, CH₃ (OEt)]. — $\text{C}_{16}\text{H}_{21}\text{NOS}_2$ (307.5): calcd. C 62.50, H 6.88, N 4.56; found C 62.67, H 6.49, N 4.38.

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[3] The compounds are racemates. Only one enantiomer is shown.

[4] The crude products were analyzed by ^1H -NMR spectroscopy (60 MHz) in every case. The isomeric purity was checked based on the integrated peak areas of the well separated 4-H signals.

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